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# **Organic Materials Chemistry**

**Date: 7 Mar 2013**

**Charles Lee  
Program Officer  
AFOSR/RTD**

**Air Force Research Laboratory**

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# 2013 AFOSR SPRING REVIEW



NAME: **Charles Lee**

## BRIEF DESCRIPTION OF PORTFOLIO:

To exploit the uniqueness of **organic/polymeric** materials technologies for enabling future capabilities currently unavailable by discovering and improving their unique properties and processing characteristics

## LIST SUB-AREAS IN PORTFOLIO:

Photonic Polymers/Organics  
Electronic Polymers/Organics  
Novel Properties Polymers/Organics  
NanoTechnology



# Organic Materials Chemistry

## Research Objective and Challenges



To exploit the u  
for en

Technologies

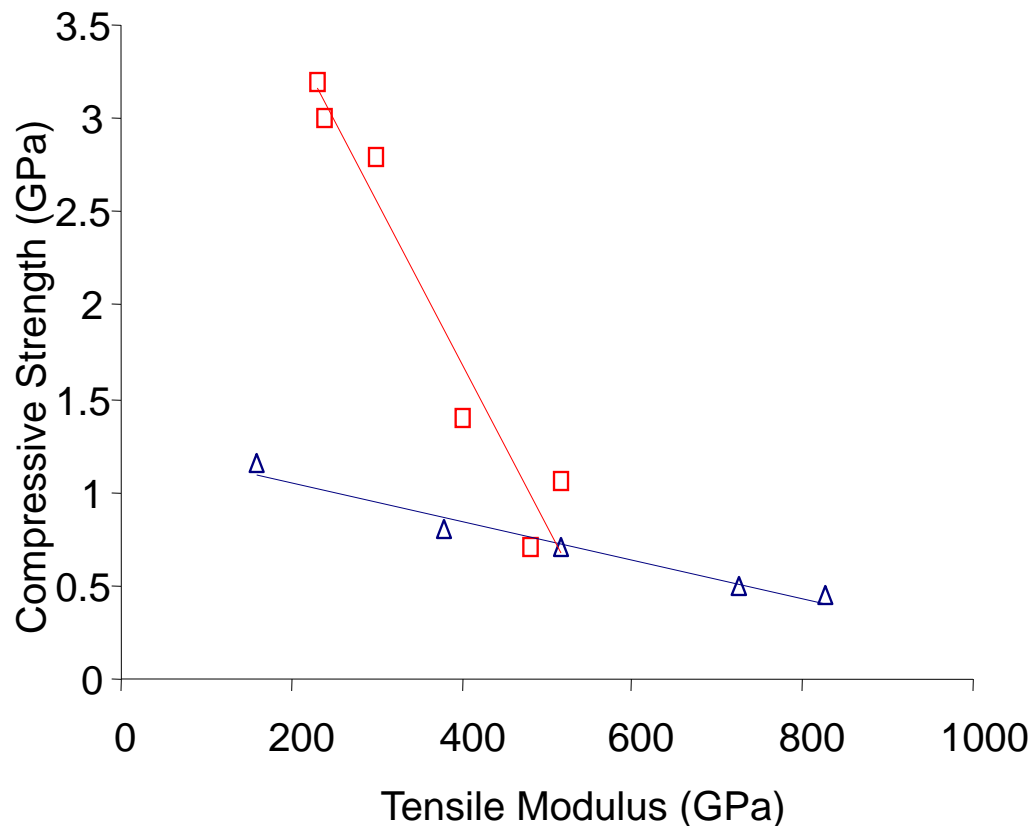
and

Challen

- Disco
- Cont
- Balar

Approa

- Molec
- Proce
- Struc



△ Pitch Based Carbon Fibers

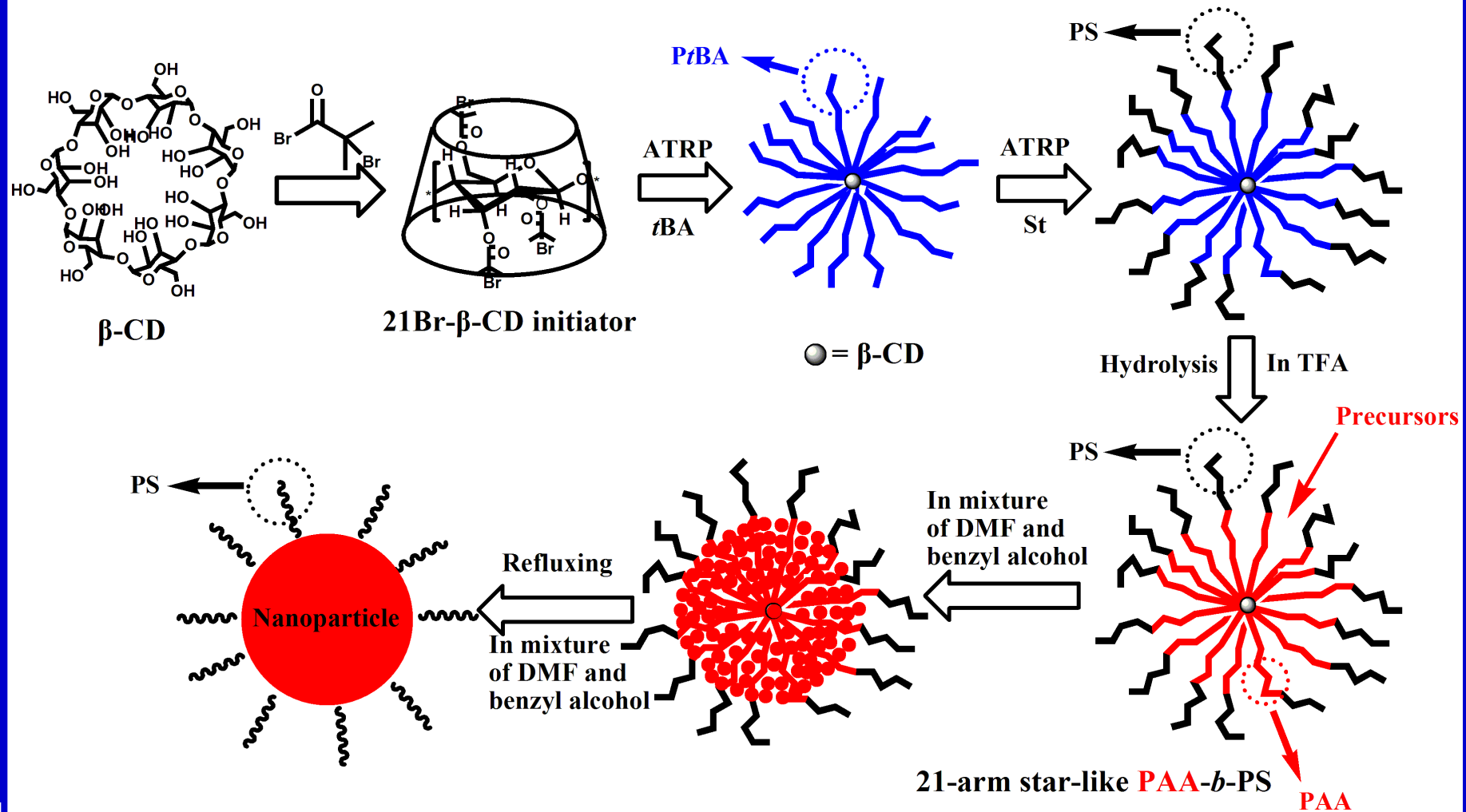
□ PAN Based Carbon Fibers

• Pro

• Not applications specific, but often use applications to guide the properties focuses

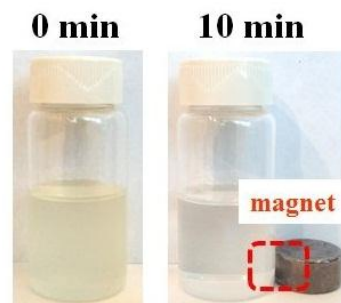
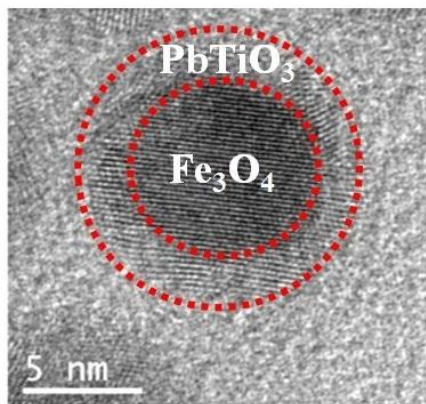
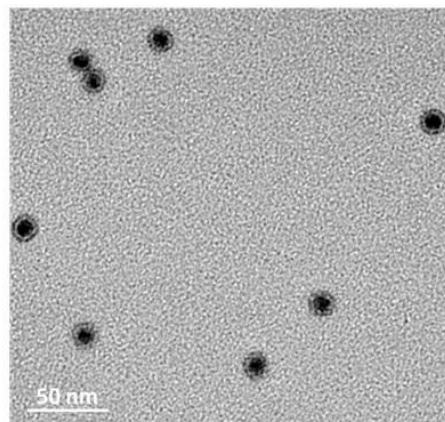
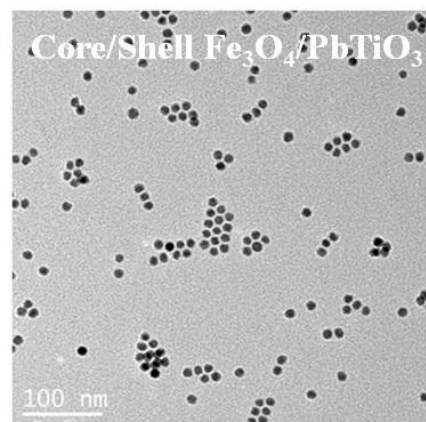
# Self Assembled Micelle vs Covalently Bonded

	Small Molecule	Block CoPolymer	Star-Like Molecule
<b>Au – Diameter (nm)</b>	<b><math>9 \pm 0.44</math></b>	<b><math>13 \pm 2</math></b>	<b><math>10.1 \pm 0.3</math></b>
<b>Grams/L</b>	<b>5.11</b>	<b>0.56</b>	<b>20.2</b>
<b># Particles/L</b>	<b><math>6.9 \times 10^{17}</math></b>	<b><math>2.5 \times 10^{16}</math></b>	<b><math>2.0 \times 10^{18}</math></b>
<b>Pt - Diameter (nm)</b>	<b><math>73 \pm 5.74</math></b>	<b><math>6.0 \pm 0.98</math></b>	<b><math>6.2 \pm 0.2</math></b>
<b>Grams/L</b>	<b>4.86</b>	<b>0.86</b>	<b>26.3</b>
<b># Particles/L</b>	<b><math>1.1 \times 10^{15}</math></b>	<b><math>3.6 \times 10^{17}</math></b>	<b><math>1.1 \times 10^{19}</math></b>
<b>Fe<sub>2</sub>O<sub>3</sub>–Diameter (nm)</b>	<b><math>16 \pm 1.49</math></b>	<b><math>10.8 \pm 2.98</math></b>	<b><math>10.1 \pm 0.5</math></b>
<b>Grams/L</b>	<b>2.94</b>	<b>1.81</b>	<b>36.2</b>
<b># Particles/L</b>	<b><math>2.6 \times 10^{17}</math></b>	<b><math>6.552 \times 10^{17}</math></b>	<b><math>1.3 \times 10^{19}</math></b>
<b>Cd-Se-Diameter (nm)</b>	<b><math>8.5 \pm 0.65</math></b>	<b>-----</b>	<b><math>9.9 \pm 0.3</math></b>
<b>Grams/L</b>	<b>0.98</b>	<b>-----</b>	<b>22.8</b>
<b># Particles/L</b>	<b><math>5.2 \times 10^{17}</math></b>	<b>-----</b>	<b><math>7.5 \times 10^{18}</math></b>
<b>PbTiO<sub>3</sub>-Diameter (nm)</b>	<b>-----</b>	<b><math>50 \pm 4.9</math></b>	<b><math>9.7 \pm 0.4</math></b>
<b>Grams/L</b>	<b>-----</b>	<b>2.12</b>	<b>31.2</b>
<b># Particles/L</b>	<b>-----</b>	<b><math>4.1 \times 10^{15}</math></b>	<b><math>7.5 \times 10^{18}</math></b>





# Core/Shell Nanoparticles – with Large Lattice Mismatch



Core/shell nanostructures are *conventionally* obtained by dissimilar materials epitaxy, which **requires moderate lattice mismatches (<2%)** between the two different materials in order to obtain high-quality core/shell heterostructures, which would otherwise be difficult to obtain.



$$D_{\text{Fe}_3\text{O}_4} = 6.1 \pm 0.3 \text{ nm (core)}$$

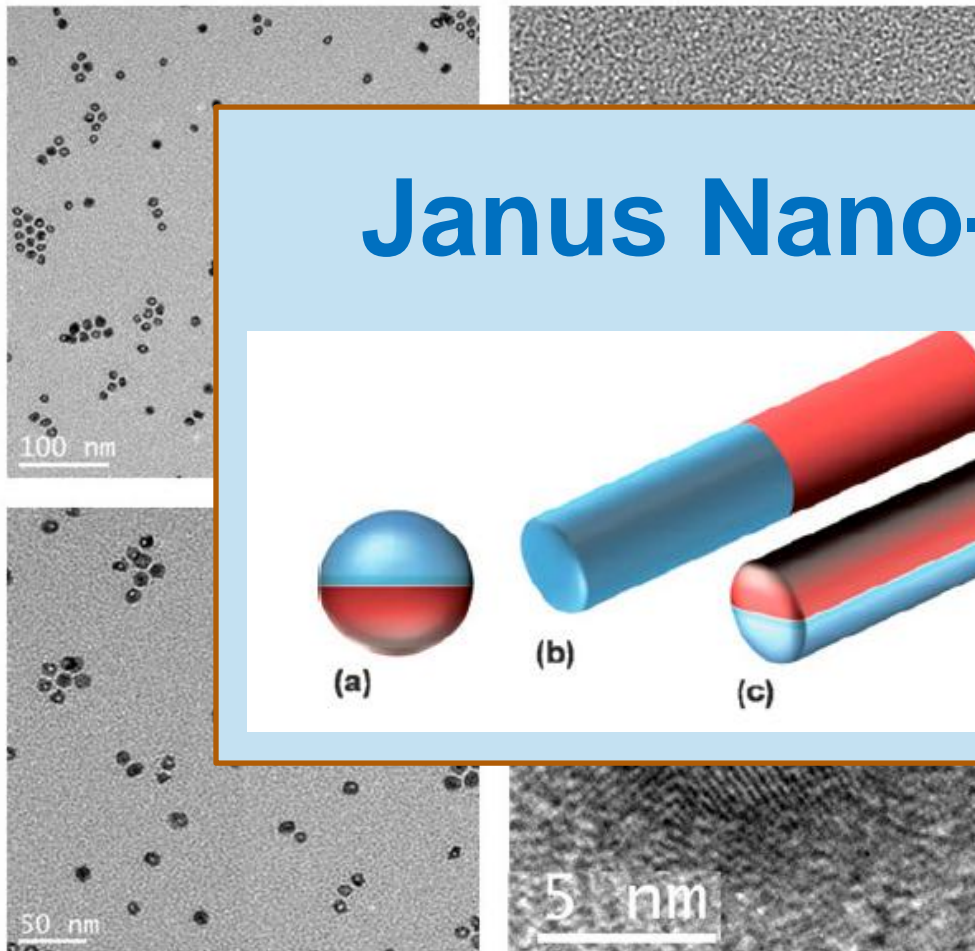
$$D_{\text{PbTiO}_3} = 3.1 \pm 0.3 \text{ nm (shell)}$$

➤ Despite more than 40% lattice mismatch between  $\text{Fe}_3\text{O}_4$  and  $\text{PbTiO}_3$ ,  $\text{Fe}_3\text{O}_4/\text{PbTiO}_3$  core/shell nanoparticles can be readily crafted by this approach!!!





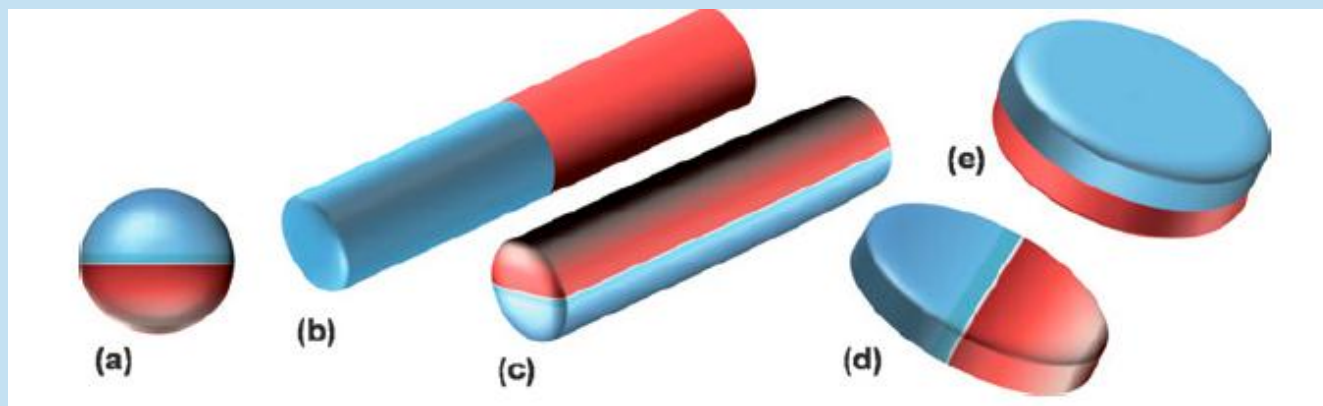
# Hollow Nanoparticles – Au Nanoparticles



Hollow noble metal

ect of

## Janus Nano-Particles



The diameter of hollow core  
 $= 5.6 \pm 0.4 \text{ nm}$





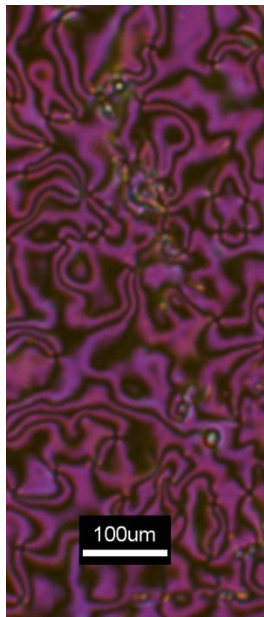
# Phototropic liquid crystals

Tim White, Tim Bunning, AFRL/RX



**“Phototropism”**: A term used to describe light induced phase changes in liquid crystals.

An example of



**Scheme for Light Induced Order-Disorder in Azobenzene Liquid Crystals**

Ikeda, J. Photochem. Photobio., 2004.

DISTRIBUTION STATEMENT A - Unclassified, Unlimited Distribution

AFRL 3

“Negative” phototropism – S (order parameter) decreases with light



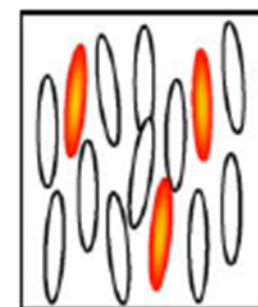
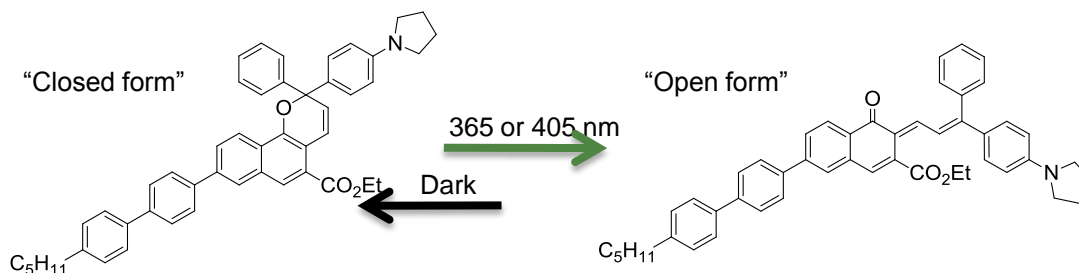
# Light Induced Disorder-Order in Naphthopyran (AM15)/LC Mixtures



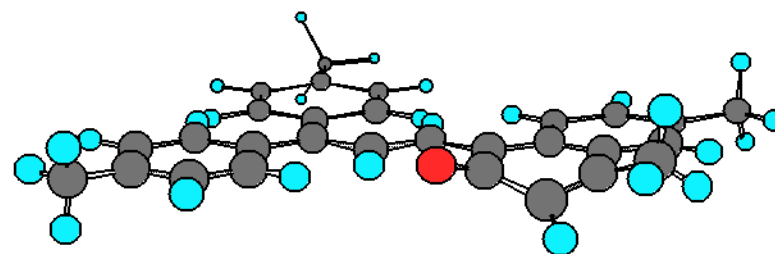
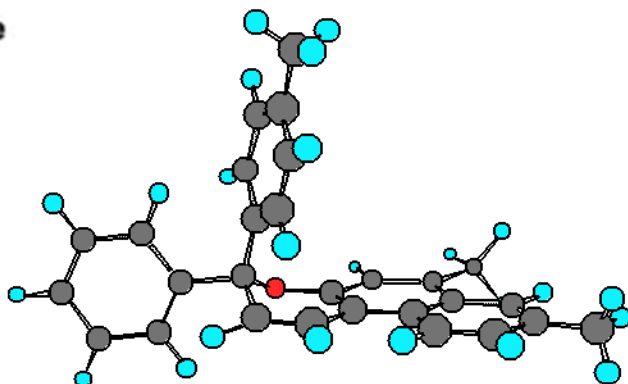
- New class of photochromic molecules that increase order upon light exposure employed for **disorder-order transitions**.
- **Demonstration of full gamut of Light Induced Phase Transitions**



I phase



N phase

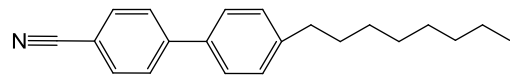


In this case of “positive” phototropism, illumination increases the compatibility of the naphthopyran as the molecular shape becomes planar and quasi-rod like aligning favorably with the liquid crystalline phases.



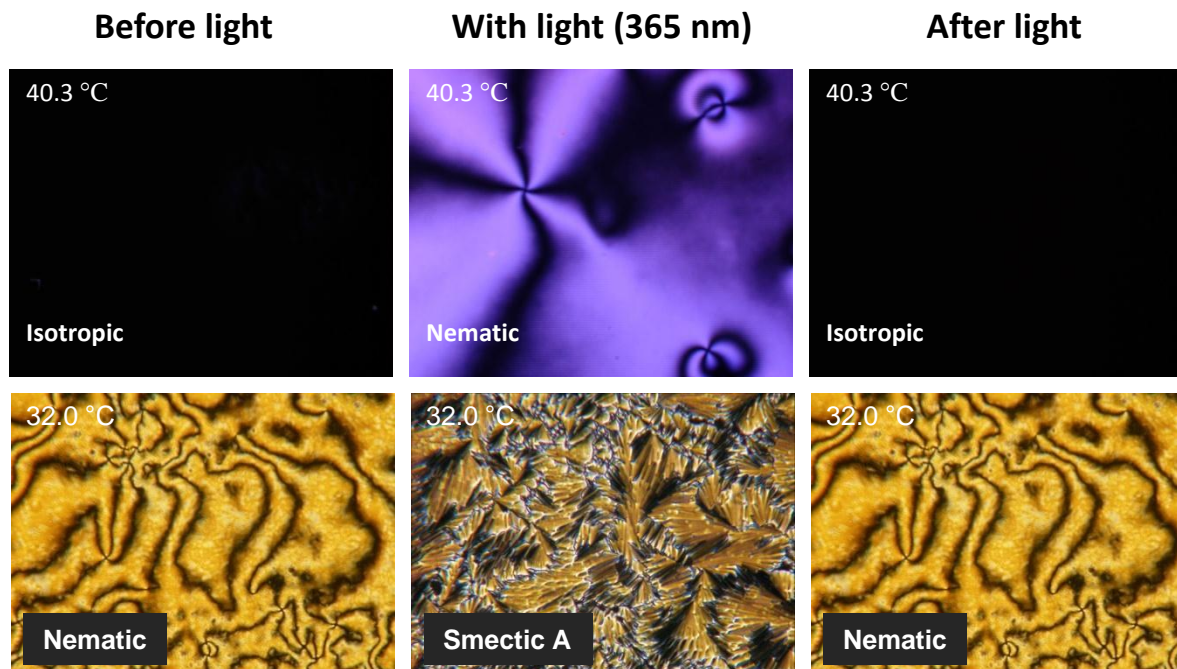
# AMI15/8CB Mixture

## Shows Additional Transition



8CB

- AMI15/8CB shows Photoinduced**
- Isotropic to Nematic Transition**
- Nematic to Smectic A Transition**



“Positive” phototropism – S (order parameter) increases with light

T. Kosa, L. Sukhomlinova, L. Su, B. Taheri, T.J. White, and T.J. Bunning, “Light Induced Liquid Crystallinity”, Nature, 2012, 485, 347-349.

DISTRIBUTION STATEMENT A – Unclassified, Unlimited Distribution

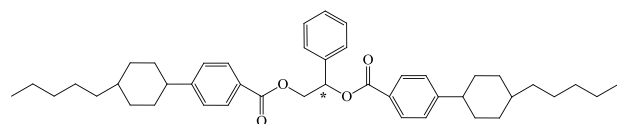


# Different Phase Change with Chiral Dopant

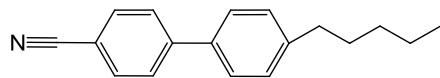


AMI15/5CB/R1011 Mixture shows Photoinduced:

- Isotropic to Cholesteric Phase Transition



R1011 – a chiral dopant from Merck

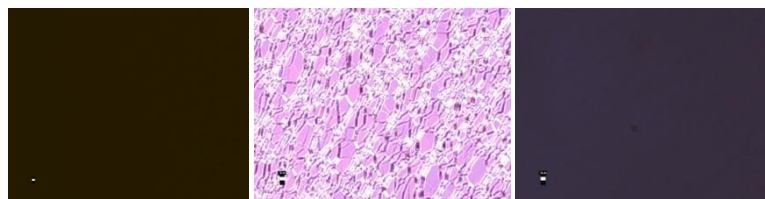


5CB

Before light

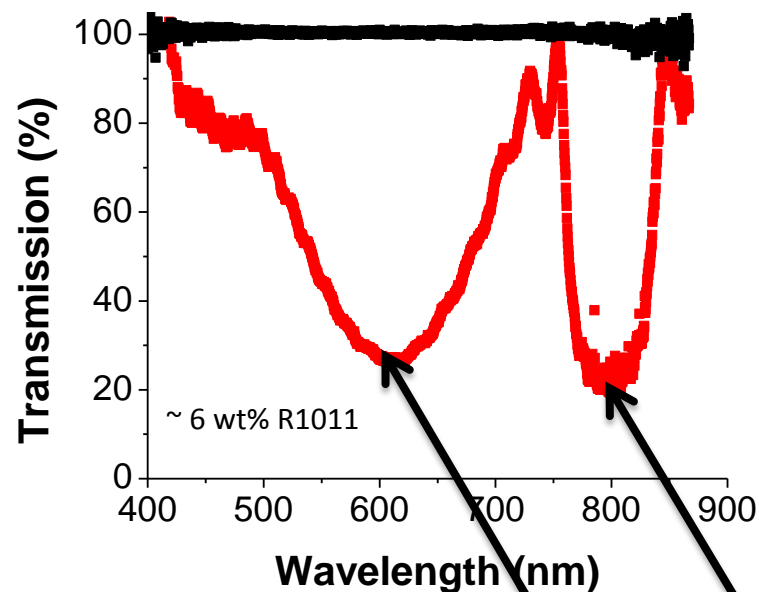
With light (365 nm)

After light



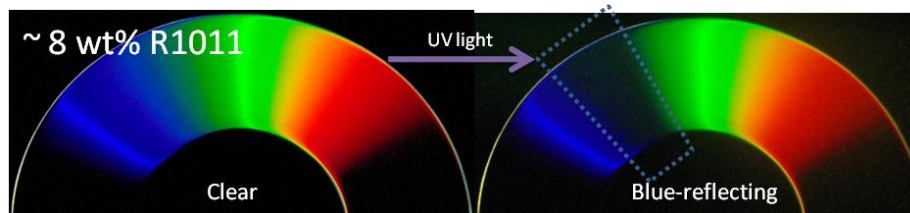
Data collected at AFRL/RX

Before irradiation – sample completely transmissive in VIS and NIR



After irradiation – sample becomes both absorptive and reflective

Data collected at AFRL/RX

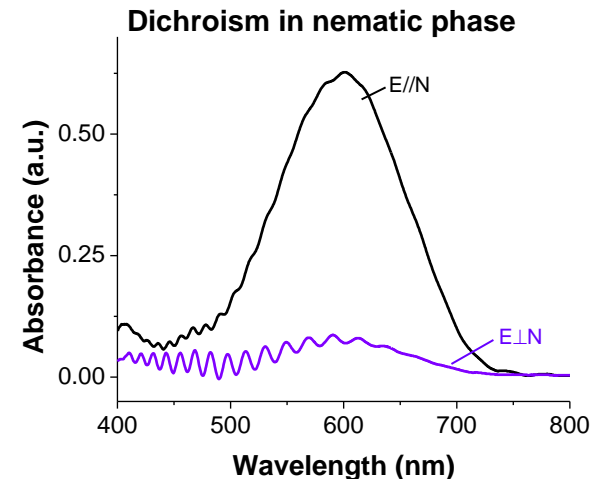
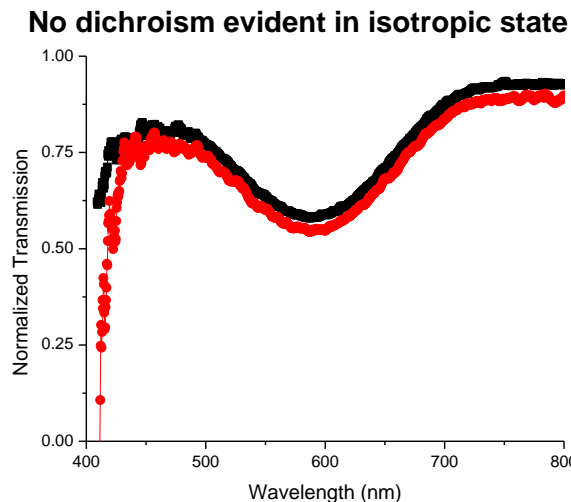




# Naphthopyran Phototropic Mixtures Unprecedented “Photo-dichroism”



**For the Isotropic to Nematic Transition in  
AMI15/5CB Mixtures,  
Dramatic light induced changes in dichroic ratio  
from  $\sim 0$  to 0.722**



**The mixture changes color and becomes polarized at the same time  
(Plain Glasses become Polarized Sunglasses)**

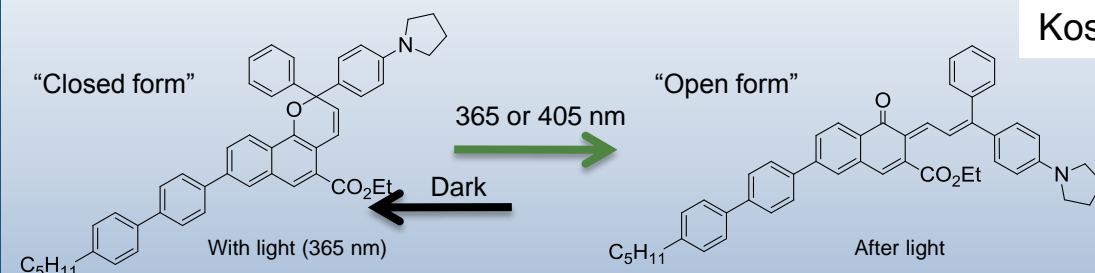
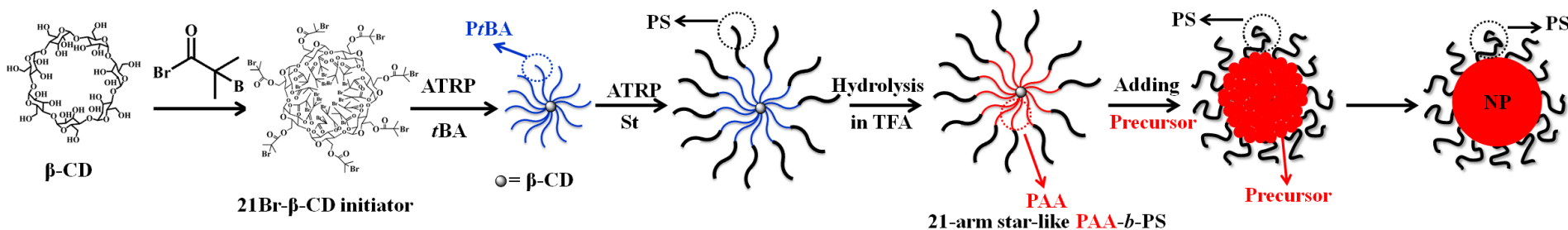




# COE Georgia Tech/AFRL Joint Project



- To craft novel *organic-inorganic nanocomposites* composed of Superparamagnetic Iron Oxide Nanoparticles (SPION) *intimately and permanently* connected with nematic liquid crystals (LCs) and chiral azo molecules with high helical twisting power (HTP) for many potential applications.



Kosa and White et. al, Nature, 2012, (485), 347–349.

**Light –induced liquid crystallinity**  
**Color switching**

potential for application in *communication devices, molecular devices, light-controllable devices, optical display system, optical data recording, photo-optical triggers, polarizers, and reflectors, and electromagnetic sensors, etc.*





# One-Dimensional Palladium Wires

Tobias Ritter (YIP), Harvard U



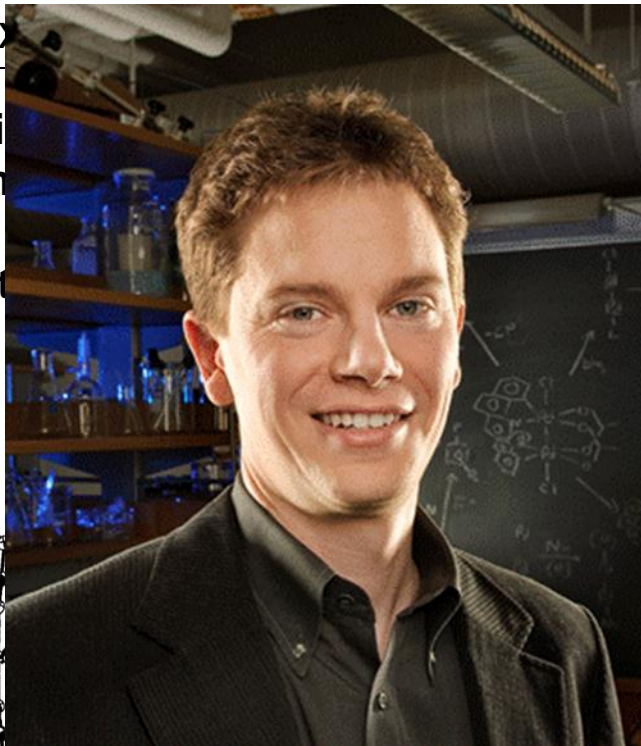
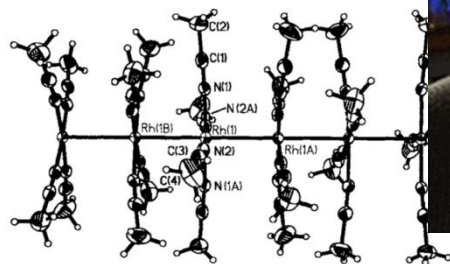
## Background on 1-D Metal Chains:

- Solid-state mixed-valence 1-D chains with Metal–Metal bonds
- Aqueous mix

-There are a few reports of infinite chains in the solid state with metal–metal bonds.

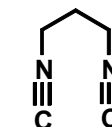
-Not solution stable; Solid-state

- take several days or weeks
- low yield (usually 50% or less)
- small scale (< 100 mg)

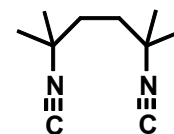


1-D metal chain  
characterized

"bridge":



"TMB":



ence ( $d^7$ – $d^8$ ) oligomers: Pt blues,  
blues, Rh oligomers.

*Chem. Ber.* **1908**, 41, 312.

*Science* **1982**, 218, 1075.

*Coord. Chem. Rev.* **1999**, 182, 263.

*Angew. Chem. Int. Ed.* **2001**, 40, 4084.

*J. Am. Chem. Soc.* **1981**, 203, 2220.

*Angew. Chem. Int. Ed.* **1969**, 8, 35.

*Angew. Chem. Int. Ed.* **1996**, 35, 2772.

*J. Organomet. Chem.* **2000**, 596, 130.

*Inorg. Chem. Commun.* **2001**, 4, 19.

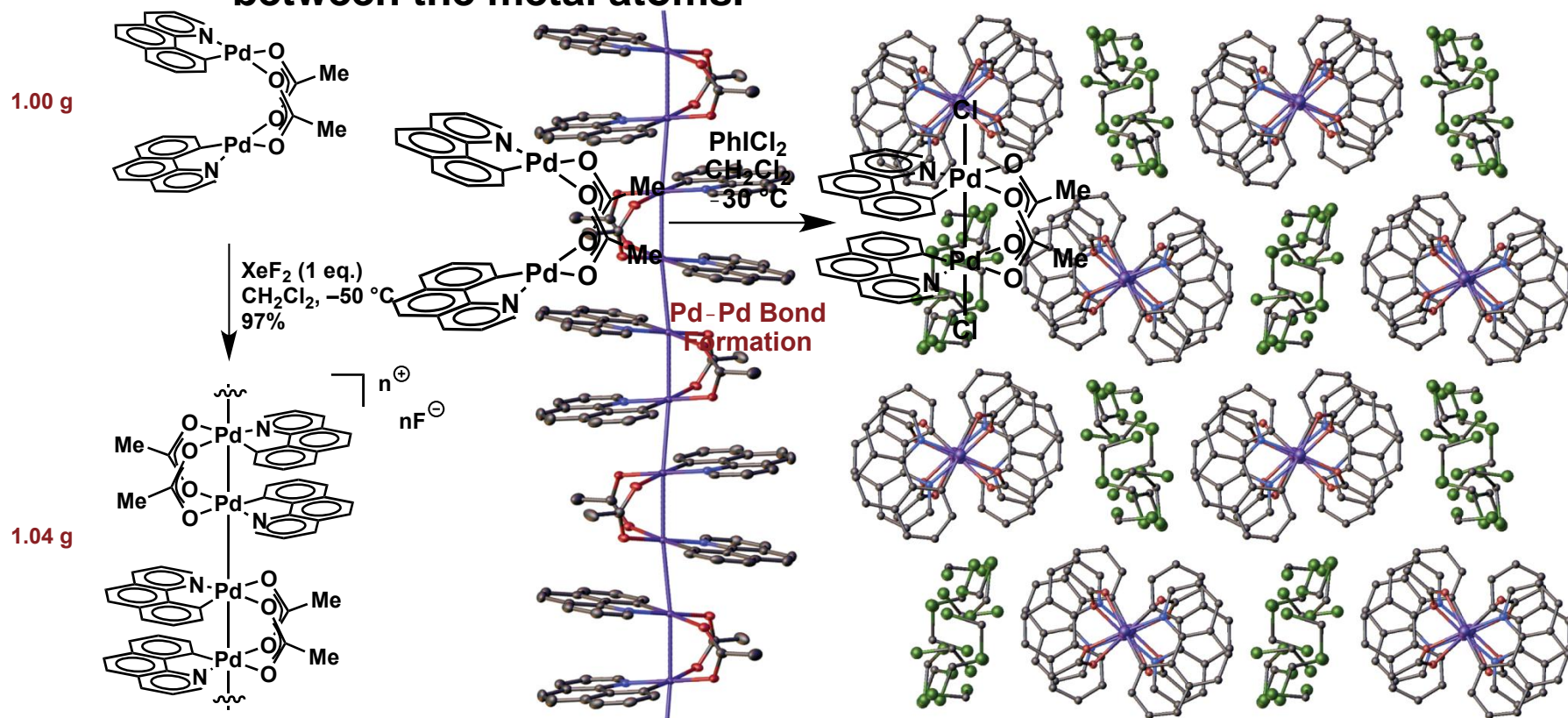


# New Chemistry – Solution Processible Palladium Wires



## From Dimers to Wires:

- Initial Pd chains in solid state revealed by X-ray crystallography
- Rapid, High-Yielding, Gram-Scale, Solution Phase Synthesis between the metal atoms.



The polymerization occurs in solution in less than 5 minutes, giving pure material on large scale

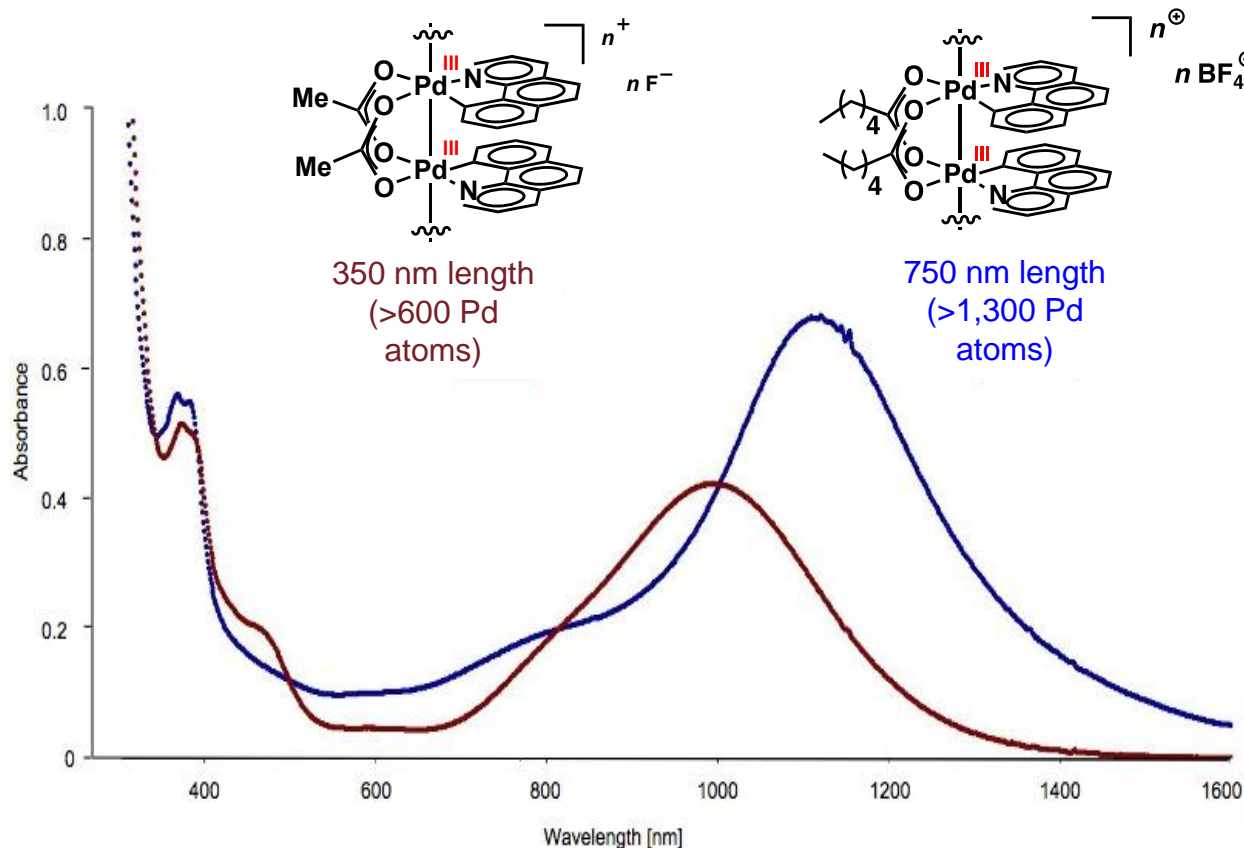


# Solution Stable 1-Dimensional Palladium Wire



## 1-D metal wires are predicted to display room temperature superconductivity

- Lengths up to 750 nm (**>1,300 Pd atoms**) observed in solution
- The longest solution-stable metal-metal bonded chain previously reported with assigned length contains **12 metal atoms**‡.
- Choice of counter-Anion controls chain length
- Enabled efficient device fabrication, not possible with previous 1-D wires



*Nature Chem.* **2011**, 3, 949–953.

‡*J. Am. Chem. Soc.* **1981**, 203, 2220–2225.





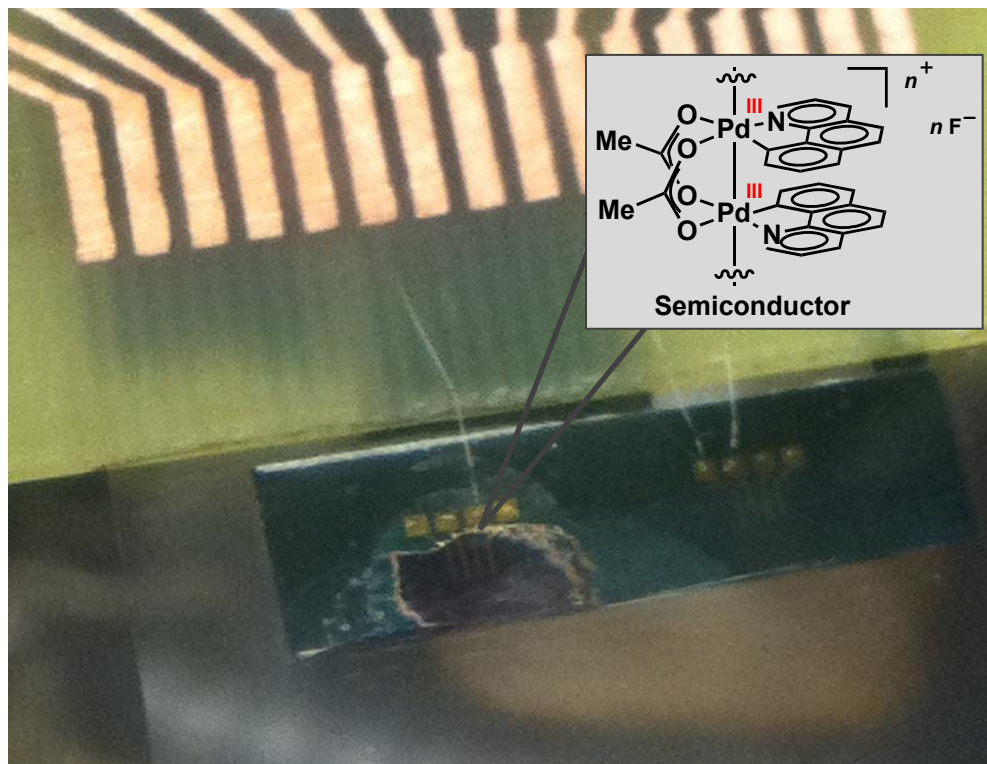
# Four Point Probe Measurement



## Thin-Film Conductivity:

- Solution processing capabilities allow for thin-film coating
- Four-point probe device used to measure conductivity of 1-D wire polymers film

Devices were fabricated using thin films of the 1-D wire polymers, which could be deposited from dichloromethane solutions either by **drop casting** or **spin coating**.



*Nature Chem.* **2011**, 3, 949–953.

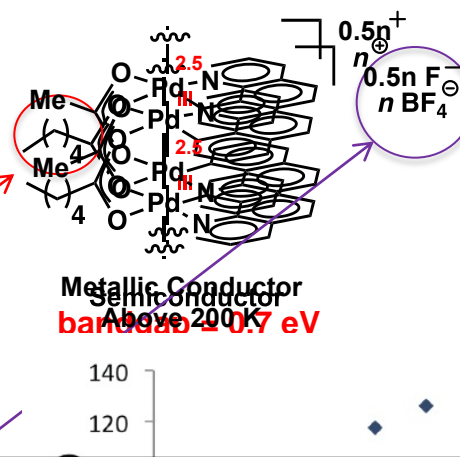
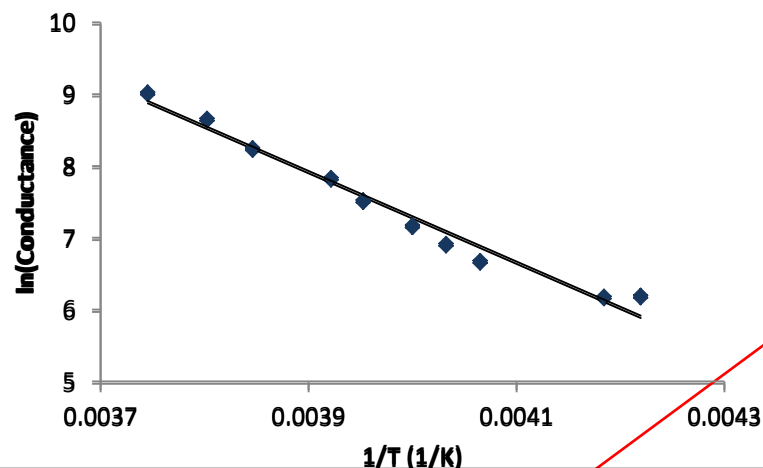


# Tuning of Electronic Properties



## Tuning Flexibility:

- Side Group Solubility
- Counter Ion
- **Pd Oxidation State**



**Films based on Pd(2.5) display the first example of a transition to a metallic state observed at ambient pressure for a polymer based on 1-D metal wires.**

Solution-stable 1-D metal wires with tunable conductive properties may have an impact on areas such as:

- Next-Generation Solar Cells
- Molecular Sensors
- Molecular Wires for Nanoscale Circuits

*Nature Chem.* **2011**, 3, 949–953

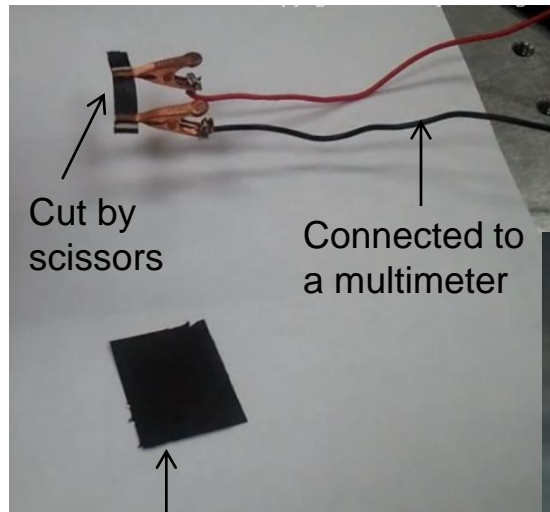


# Power Generation with Body Heat

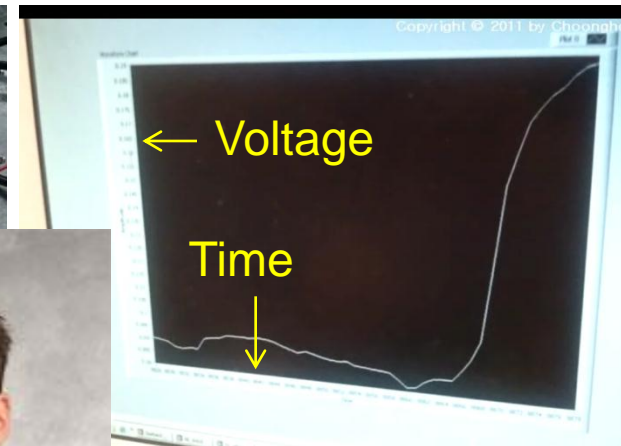
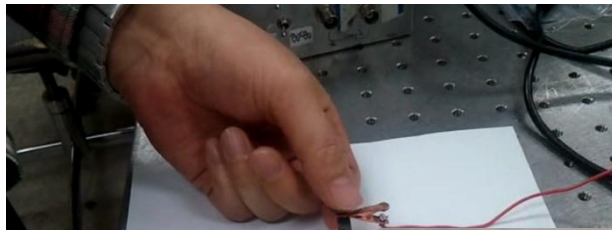
Choongho Yu & Jaime Grunlan, Texas A&M



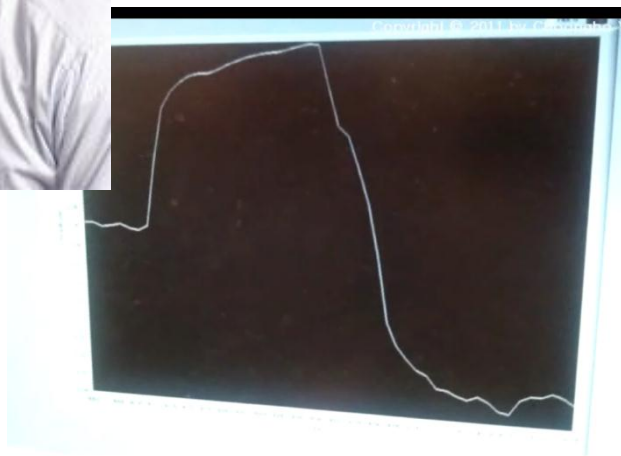
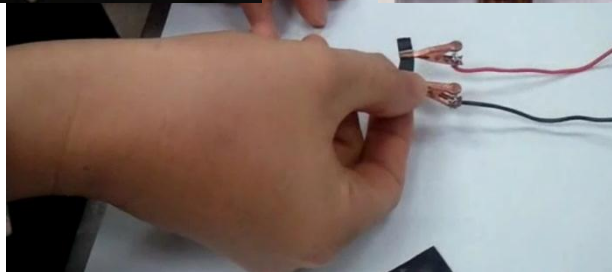
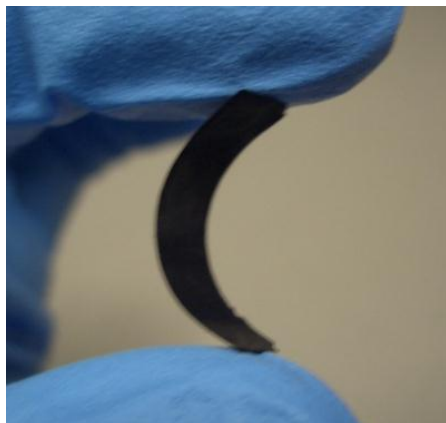
First demonstration of electricity generation from polymeric materials



Flexible TE polymers



Voltage - Time response





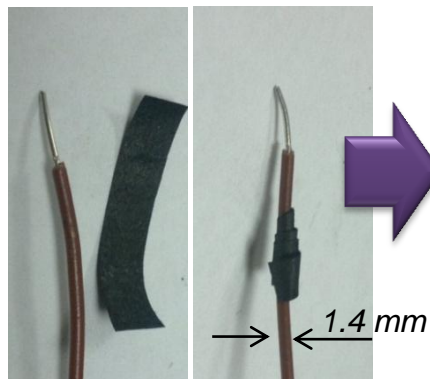


# Air-stable fabric thermoelectric modules made of n & p-type composites



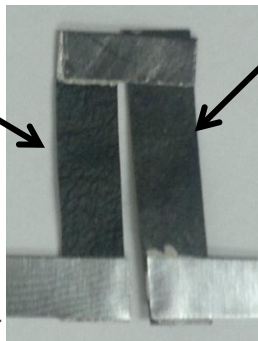
**Objective: Demonstrate power generation & cooling with organic composites**

## (1) Flexible composite



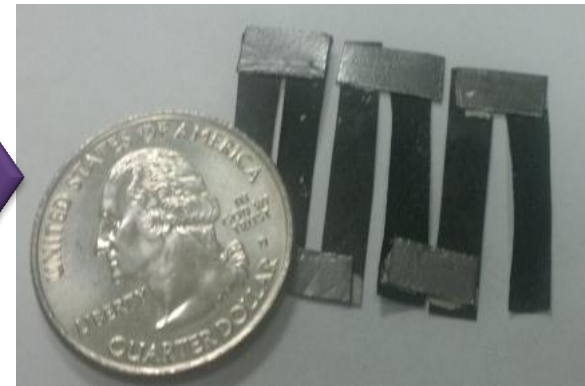
N-type  
Carbon  
nanotubes +  
Poly-  
ethyleneimine  
(PEI) +  
 $\text{NaBH}_4$  treatment

## (2) Module fabrication

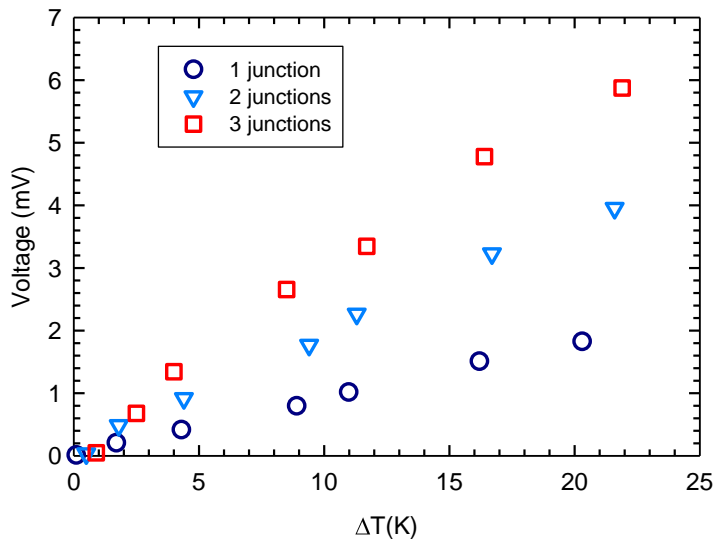


P-type  
Carbon  
nanotubes +  
Paper  
(cellulose  
fibers)

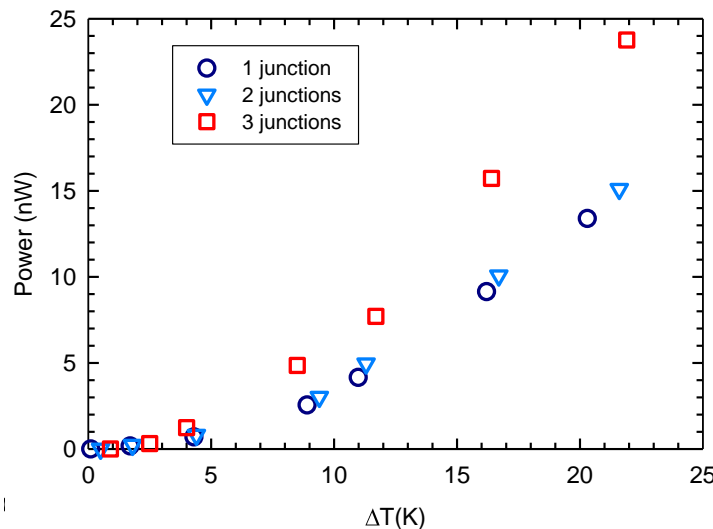
## (3) Multiple junctions in series



### Voltage output vs Temperature



### Power output vs Temperature



**Voltage and power are being increased by:**  
(a) stacking more layers;  
(b) connecting more modules



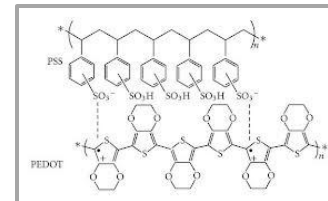
# Improving Power Factor by Tuning P-type composites with multiple CNT stabilizers



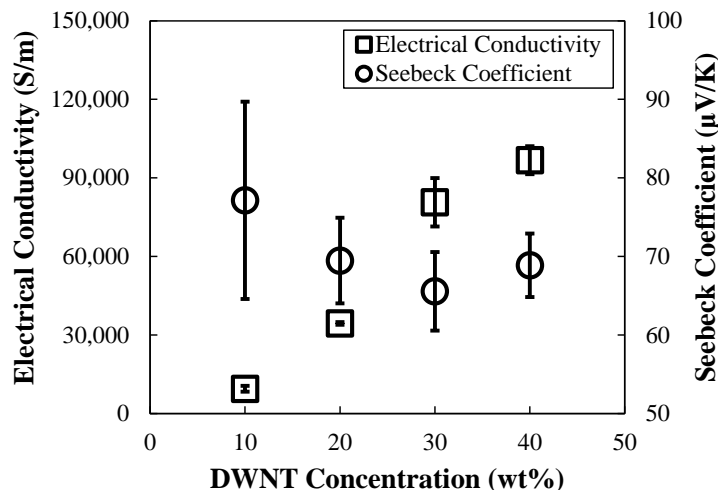
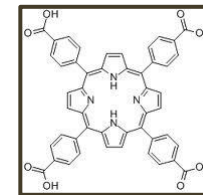
**Objective: Produce the highest possible power factor (PF) for fully organic, flexible composites**

- Double-walled carbon nanotubes (DWNT) are stabilized with two different molecules in poly(vinyl acetate) latex:

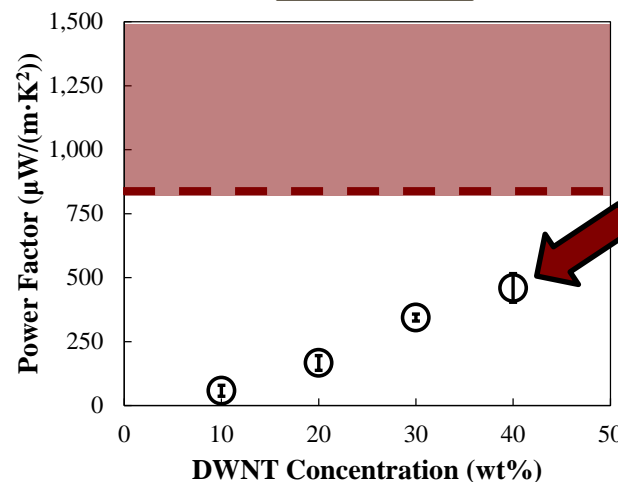
- PEDOT:PSS (conductive)**



- TCPP (semi-conductive)**



Electrical conductivity increases with DWNT concentration; while the Seebeck coefficient remains relatively insensitive.



Highest PF ever reported for fully organic composite at ~500 μW/(m·K²)!

The power factor ( $S^2\sigma$ ) increases with DWNT concentration and is within an order of magnitude of traditional inorganics (maroon shaded region).



# Different Module Design Concept

David Carroll, Wake Forest U.



Using

s and

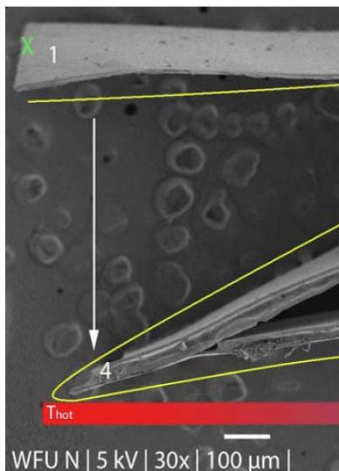
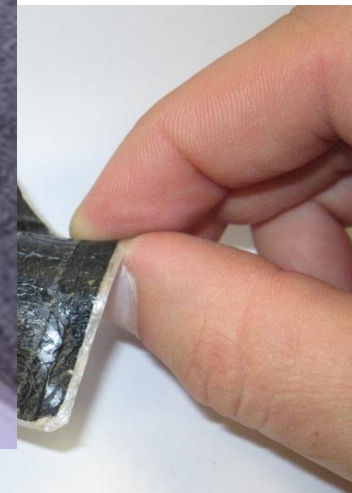


Figure 11: SEM image of multilayer films in the p-type films (1), and -V in the films (4) adds the potentials in series.



The garment has recently been shown on CNN International, CNBC, and the Discovery Channel.



# Photorefractive Polymers

## Multi-TD's Interests



- Laser Refraction
- Optical Signal Processing
- Wave Front Correction
- 3D Holographic Display
- Image Correlation



<u>Earlier Results</u>		<u>Now</u>
Luminance	350Cd/m <sup>2</sup>	1000Cd/m <sup>2</sup>
Image Holding	30s <50Cd/m <sup>2</sup>	2min >200Cd/m <sup>2</sup>
Sensitivity	200mW	1W






# Two Beam Coupling Optical Correlation

Jed Khoury AFRL/RY (11RY01COR)



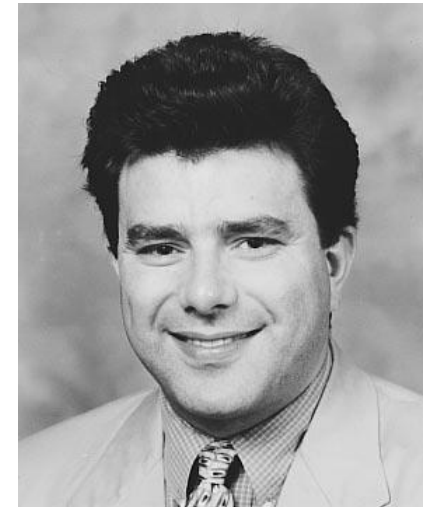
**Jed Khoury**



**Charles Woods**



**Bahareh Haji-saeed**

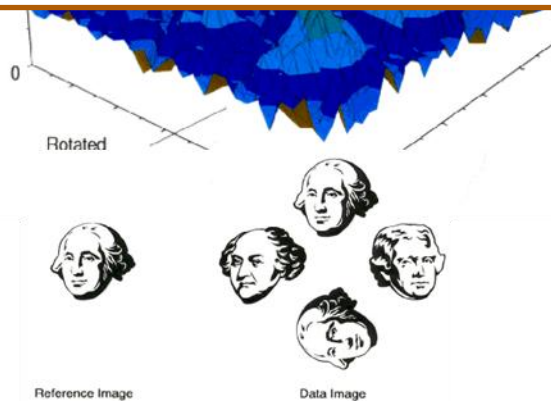


**George Asimellis**

compression developed by  
AFRL/RY (Jed Khoury)

2. Organic photorefractive material  
that was developed by University  
of Arizona/Nitto Denko

**Both efforts funded by AFOSR**

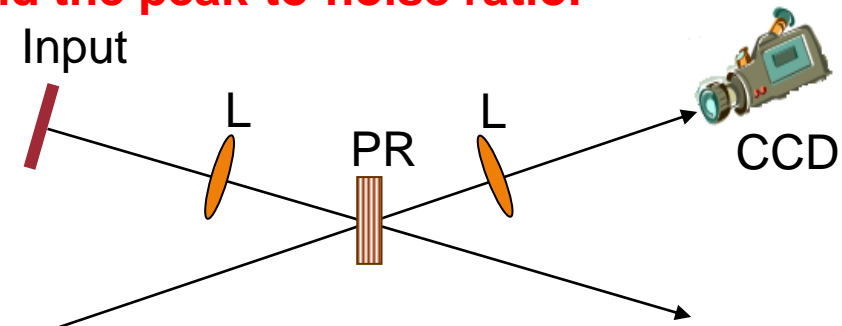
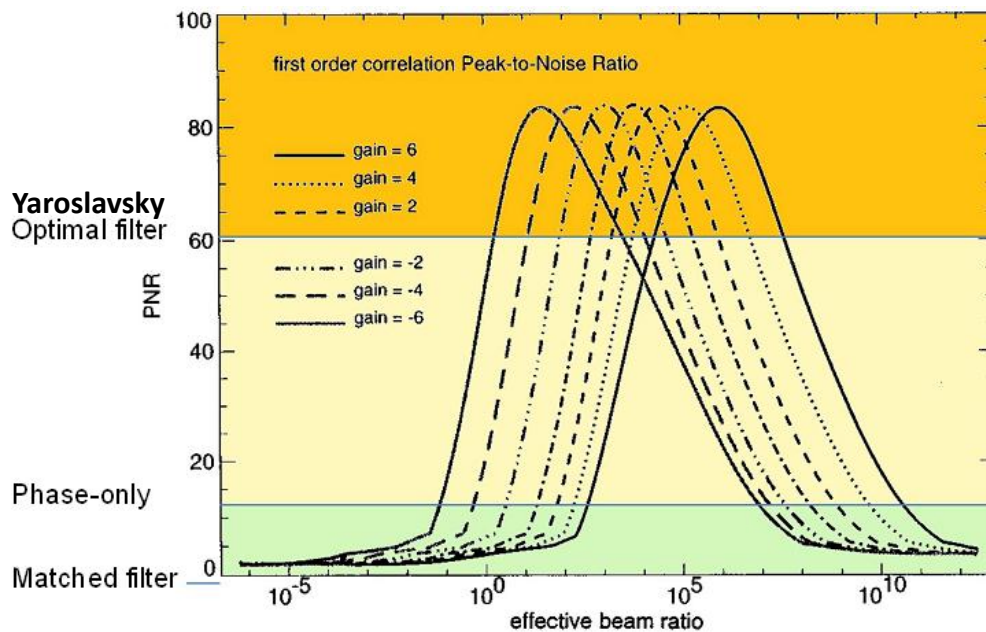




# Computer Simulation Comparing Two-Beam Coupling Correlation vs SOA Correlation Algorithms



**No correlation filter in the last 50 years, since the first correlation invented by Vander Lugt (1963), have been designed that can improve simultaneously the discrimination, the signal-to-noise ratio, and the peak-to-noise ratio.**



Using input that has a lot of background noise, Two Beam Coupling Correlation is:

- 1.5X better than Yaroslavsky Optimal filter
- 10X better than Phase-only filter
- superior to Matched filter (failed to recognize target)

**But the scheme will require very large beam ratio, that will require a photorefractive material that has very high diffraction efficiency.**



(a)

0.040

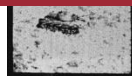
(b)

0.025

(c)

8

(d)



Input

Matched filter

Phase-only filter

Two Beam Coupling  
Compression filter







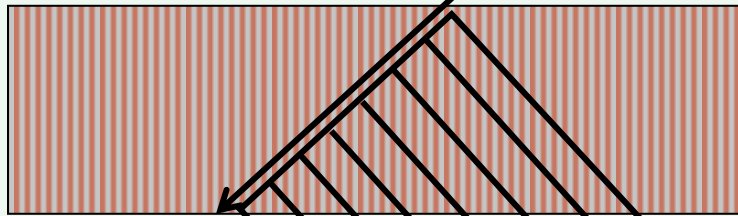
# BULK PHOTOREFRACTIVE CORRELATION VS THIN FILM PR POLYMER CORRELATION

Jed Khoury, AFRL/RV



A Thick BSO Crystal

Point source  
( $\delta$ -function input)



Thick diffracted beam  
(Broad impulse response )

A Thin Nitto Denko Organic Material

Point source  
( $\delta$ -function input)



Thin diffracted beam  
(Narrow impulse response )

Dephasing Factor is small in thin film holographic materials.

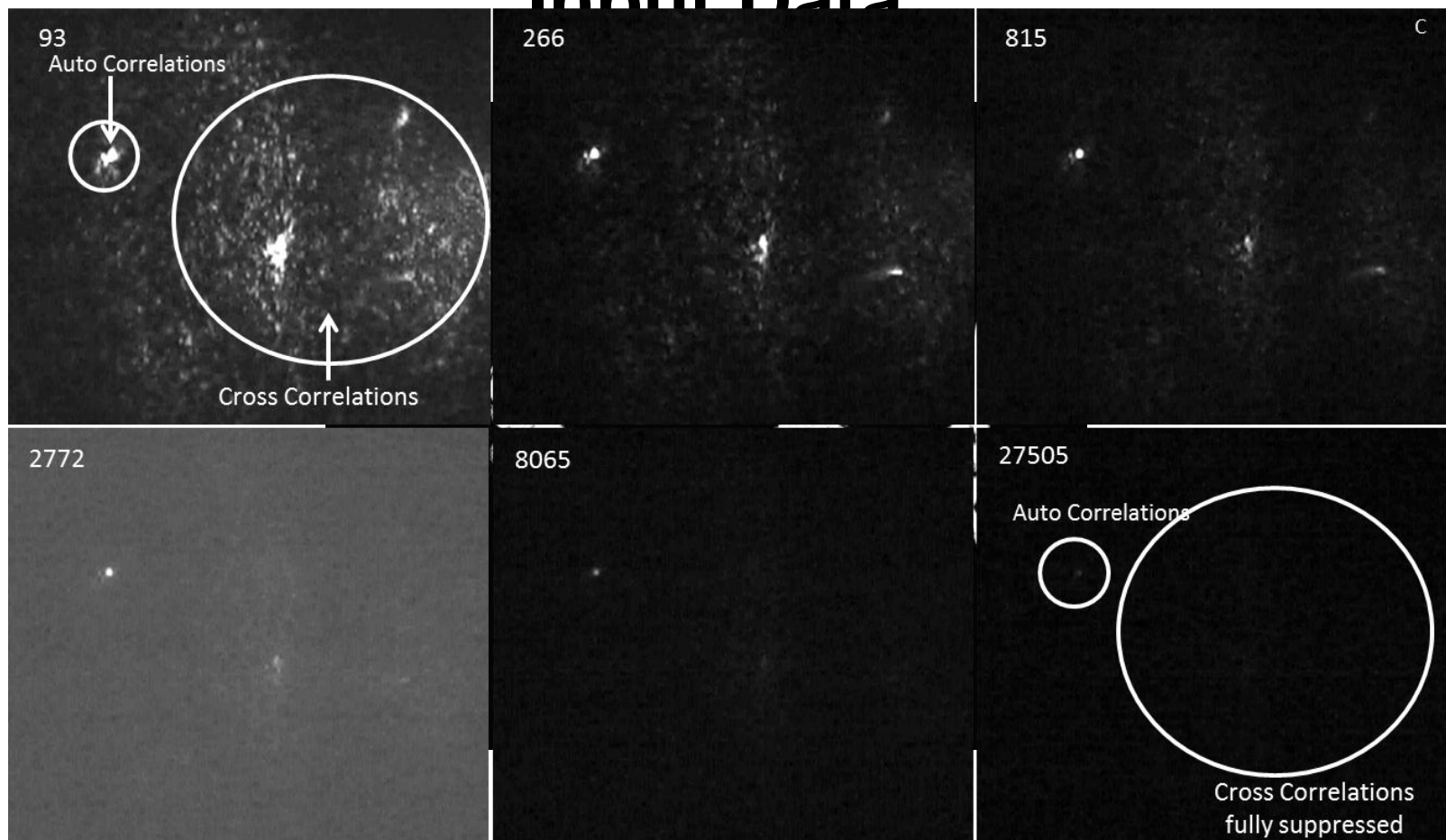


# Two Beam Coupling Experiment with PR Polymer Thin Film(1)



Dynamic range compression increases

Input Data



Dynamic range compression increases

DISTRIBUTION STATEMENT A - Unclassified, Unlimited Distribution

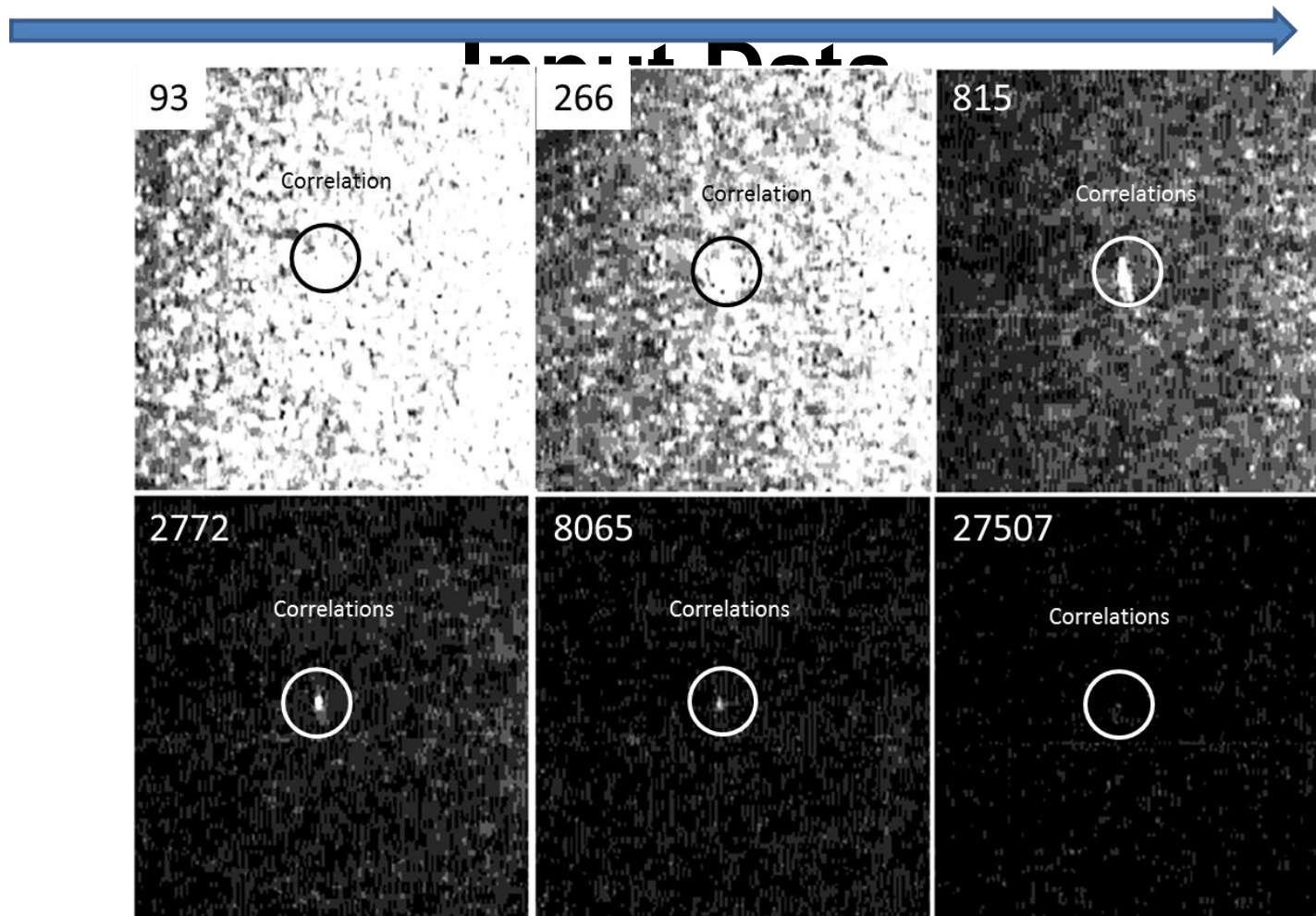




# Two Beam Coupling Experiment with PR Polymer Thin Film (2)



Dynamic range compression increases



Dynamic range compression increases

DISTRIBUTION STATEMENT A – Unclassified, Unlimited Distribution

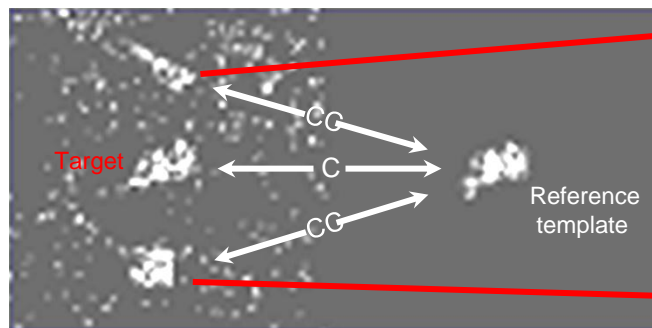




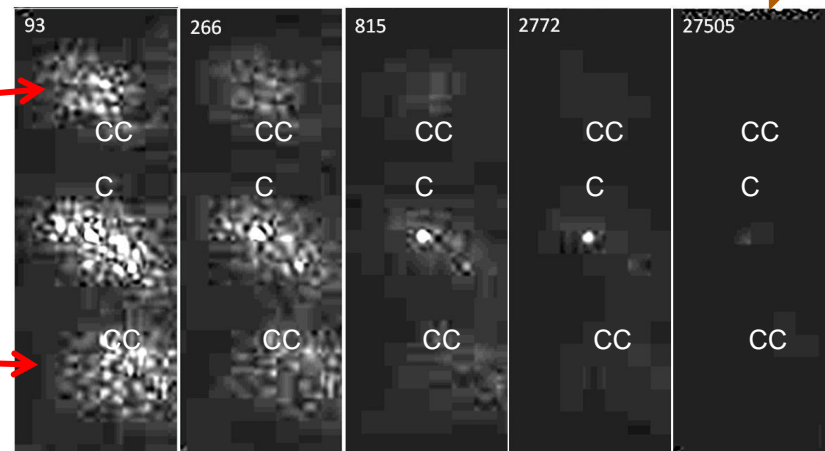
# Applied to Synthetic Aperture Radar Data



Low resolution images synthesized from the MSTAR data base



Dynamic range compression increases



The first correlation filter that can improve simultaneously the

- SNR (100X)
- PNR,
- Discrimination (3 orders of Magnitude)

**Material Chemistry Makes It Possible!!!**

Correlation filter that outperforms optimal digital correlation filters



# Portfolio Trends



## **Decreasing Emphases:**

- Organic Solar Cells**
- Organic Transistors**

## **Increasing Emphases:**

- Self Assembly in Solid State**
- Radical, Spin and Excited State Controlled Properties**

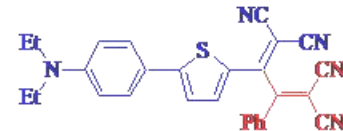
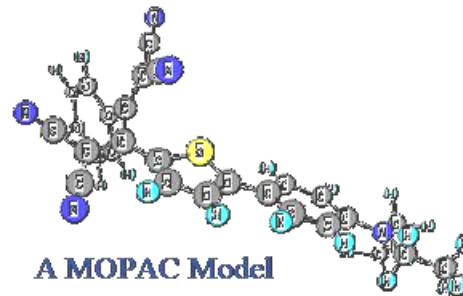




# Summary



- Program Focused on developing New and Controlled Properties
- Not applications specific, but often use applications to guide the properties focuses
- Scientific Challenges
  - Discover New Properties
  - Control Properties
  - Balance Secondary Properties
- General Approaches
  - Molecular Design
  - Processing Control
  - Establish Structure Properties Relationship



Flexible Photodetector

